

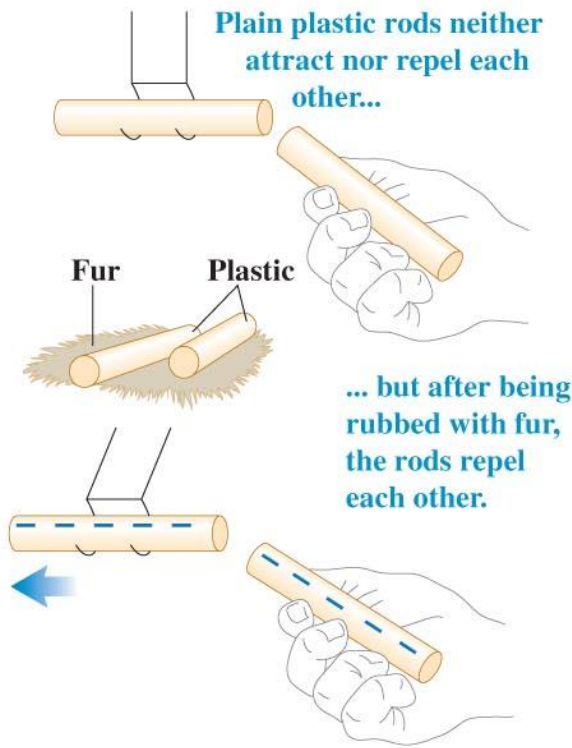
# Chapter 16

Electric Charge and Electric Field

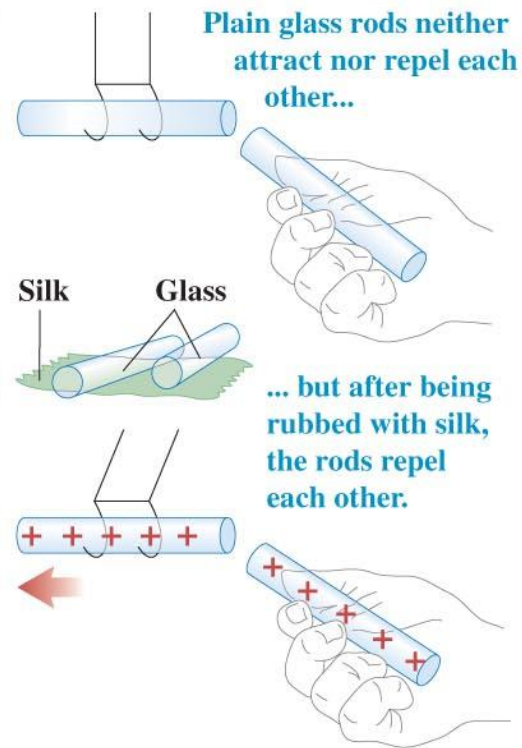
# Electric Charge: Why Important

- Many natural phenomena (static shock, lightning,...)
- Huge application in the distribution and use of electricity (not really electrostatic)
- Ultimately responsible for the interactions between matter (other than gravity)

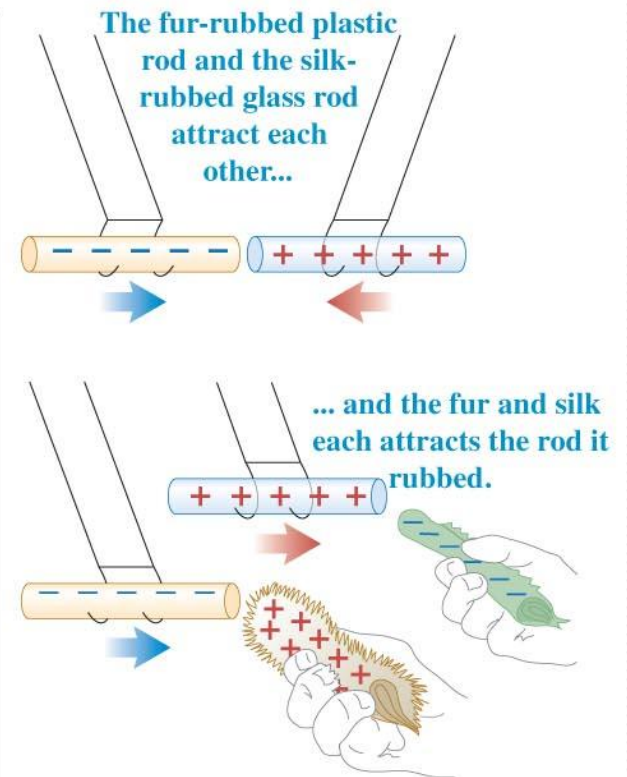
# Static Electricity



**(a) Interaction between plastic rods rubbed on fur**



**(b) Interaction between glass rods rubbed on silk**



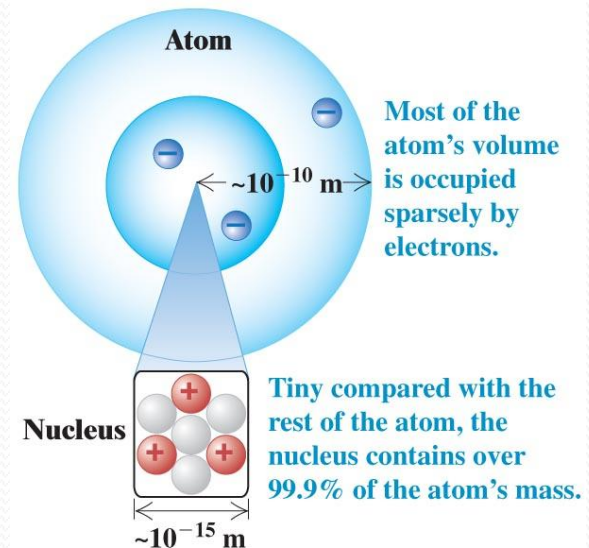
**(c) Interaction between objects with opposite charges**

# Properties of Electric Charges

- There are only two types of electric charge: positive and negative
  - like charges attract, opposite charges repel
- Charges are neither created nor destroyed – they are redistributed
  - the net amount of electric charge produced in any process is zero

# Electric Charge in the Atom

- Electron:  $-e$
- Proton:  $+e$
- Neutron: no charge
- Atoms: nucleus of protons/neutrons surrounded by electrons
- Atoms are neutral, so the number of protons = the number of electrons



**+** **Proton:** Positive charge  
Mass =  $1.673 \times 10^{-27}$  kg

**○** **Neutron:** No charge  
Mass =  $1.675 \times 10^{-27}$  kg

**-** **Electron:** Negative charge  
Mass =  $9.109 \times 10^{-31}$  kg

The charges of the electron and proton are equal in magnitude.

# Elementary Charge

- Electrons and protons have charge  $e=1.602\times 10^{-19}$  C
  - by convention, the charge of the electron is negative
- Electric charge is quantized
  - any amount of charge must be a multiple of  $e$  (unless we want to discuss quarks, that is)
- A neutral particle can decay into charged particles ( $n\rightarrow p+e$ ) and two particles can annihilate their charges ( $e^++e^-\rightarrow\gamma\gamma$ ), but the total amount of charge in the system never changes

# Insulators and Conductors

- Some materials allow charges to move freely within them → conductors
  - examples: metals, salt water
  - in metals, electrons are nearly free and can easily move from one atom to another
  - in salt water, molecules of salt (NaCl) are dissolved into charged ions which move freely in the liquid
- In other materials, charges cannot move or can only move with great difficulty → insulators
  - examples: glass, pure water
- There are semiconductors, too

# Coulomb's Law

$$F = k \frac{Q_1 Q_2}{r^2}$$

2.000000000000000000...

$$k = 8.988 \times 10^9 \text{ N} \cdot \text{m}^2 / \text{C}^2$$

$$k = \frac{1}{4\pi\epsilon_0}$$

$\epsilon_0 = 8.85 \times 10^{-12} \text{ C}^2 / \text{N} \cdot \text{m}^2$

“permittivity of free space”

denominator!

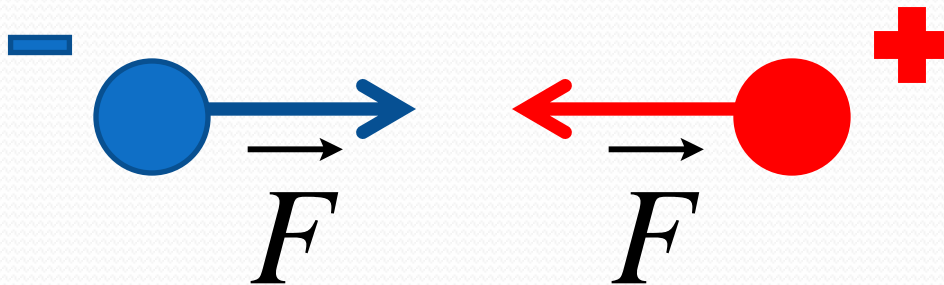


# Electric Force is a Vector

... like any other force

trivial if there are two charges

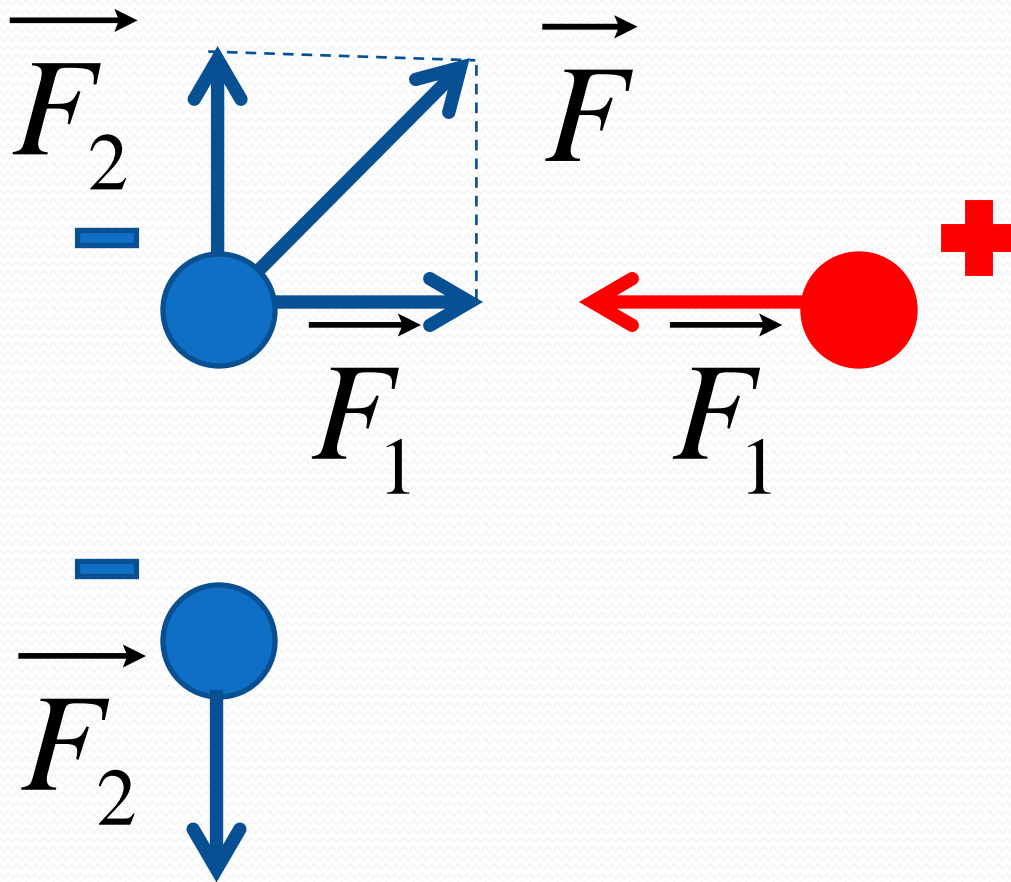
what if there are more than two?



# Adding Electric Forces

- Everybody is affected by everybody else
  - if there are  $n$  charges, there are  $(n-1)$  forces acting on each charge
- Like charges repel, opposite charges attract
- When solving a problem, first figure out the number of dimensions
- Remember: forces are vectors!
- Cross check: sum of all forces acting on all charges is zero

# Vector Sum of Forces



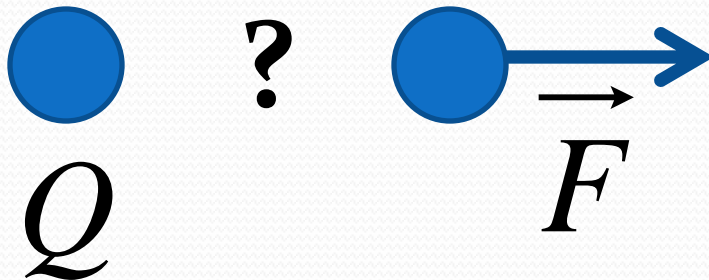
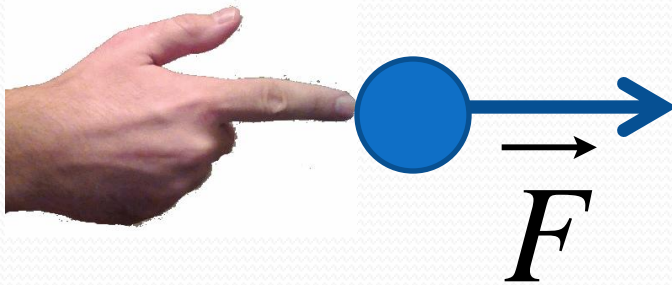
$$\vec{F} = \vec{F}_1 + \vec{F}_2$$

$$F_x = F_{1x} + F_{2x}$$

$$F_y = F_{1y} + F_{2y}$$

# Electric Field

- How come that one electric charge pushes (or pulls) another electric charge without touching it?



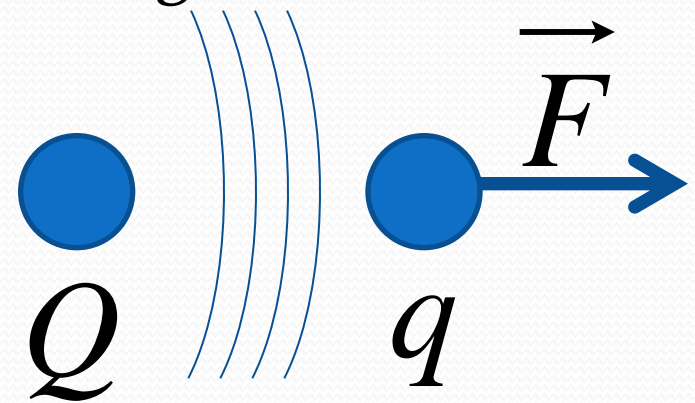
The answer (M. Faraday):  
there is **field**

- It extends outwards the charge and penetrates the whole space
- Fields from multiple charges add up (“superposition”)

# Electric Field and Electric Force

- Electric field is a property of the charge that emits it, but does not depend on the probe charge

$$E = \frac{F}{q} = k \frac{Q}{r^2}$$



- Electric field is a vector

$$\vec{E} = \frac{\vec{F}}{q}$$

two similar types of problems:

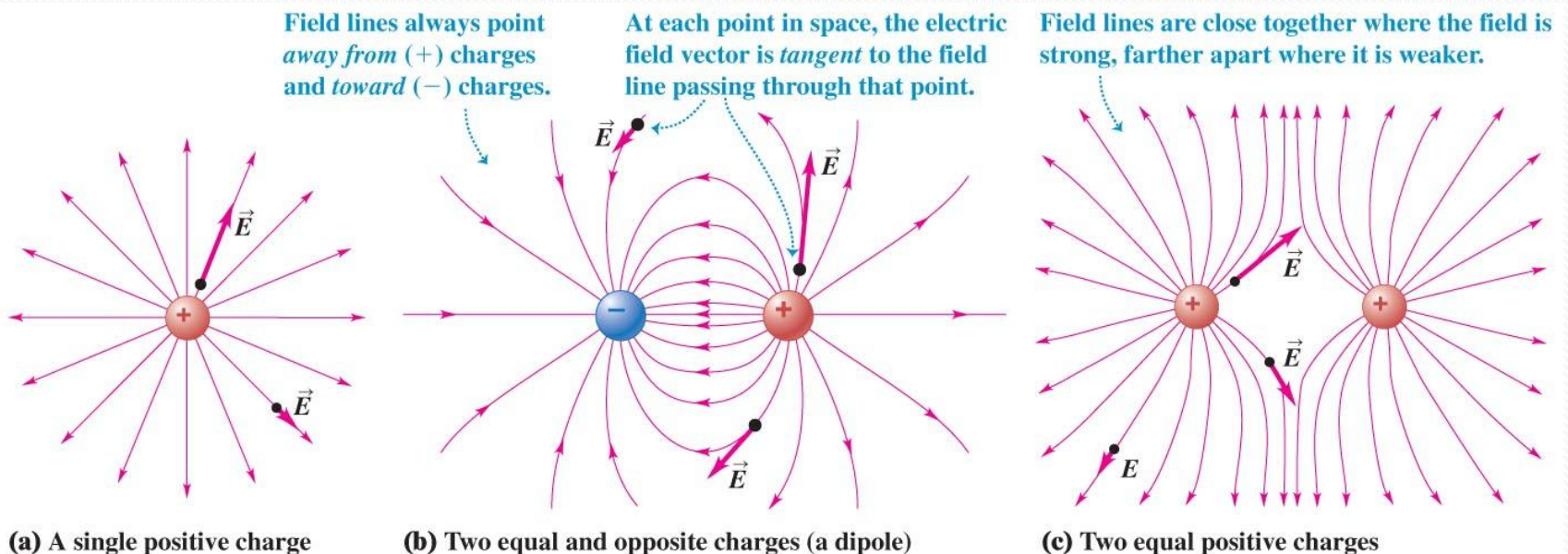
➤ find the force acting on a charge

➤ find electric field at a given point

make sure you understand the question!

# Field Lines

- Field lines are drawn so that they indicate the direction of the force on a positive probe charge
- More on them when we'll talk about potential



# Electric Fields and Conductors

- In electrostatics, the electric field inside a conductor is zero
  - otherwise the charges would experience a force and move – not static anymore
- Outside the conductor, the electric field is always perpendicular to the conductor surface
  - otherwise the charges would move along the surface
- Electric field inside a hollow metal box is zero
  - a person inside such a box (e.g. in a car) is protected against a strong electric discharge (e.g. lightning)